## **Sooting Flames # 5**

# **Apparatus**

Premixed flames of *ethylene/oxygen* at *atmospheric pressure* were produced on a commercial McKenna sintered bronze burner (d=60mm). The burner was water-cooled and the temperature of the cooling water was kept constant at 60°C. An external shield of nitrogen was used to avoid air entrainment. Different cold-gas flow velocities and equivalence ratios were used.

## Measurements

Flame temperatures were measured along the flame axis with a fast-response thermocouple (silica-coated 25  $\mu$ m Pt/Pt-13%Rh) by using a fast-insertion procedure. A radiation correction procedure was applied to obtain corrected temperature profiles [Ref 5]. The uncertainty of the measured temperatures was estimated to be as high as 100 K.

Soot, condensable species (CS), and gaseous combustion products were isokinetically sampled along the flame axis by using a stainless-steel water-cooled (80°C) probe (i.d. = 2mm) [Ref 2,4].

Species concentrations were sampled in two gas chromatograph valves with online gas analysis, GC-TCD for small species, GC-FID for C<sub>1</sub>-C<sub>6</sub> species, and GC-MS for PAH [Ref 2, 5]

Soot and CS were collected on a teflon filter and in a cold trap and extracted by dichloromethane (DCM) to separate the DCM-soluble material (condensed species) from the insoluble solid carbonaceous material (soot). The amount of soot was determined gravimetrically [Ref 2].

H/C ratio of soot was measured by a Perkin–Elmer 2400 CHNSO elemental analyzer [Ref 7].

#### **Conditions**

Pressure: 1 bar

φ=2.4 (C/O=0.8) – Fuel: 44.4 % - O2: 55.6 %

- V<sub>0</sub>=2cm/s *Ref 2 (mostly) & 4* 
  - o Temperature profile
  - Species profiles (C2H4, C2H2, C6H6) Ref 2 & 3
  - o PAH profiles (A2, A2R5, A3, A4, A4R5, FLTN) Ref 3 & 4
  - Soot profiles
- V<sub>0</sub>=4cm/s Ref 2 (mostly), 4, 5, 6, 7
  - o Temperature profile Ref 2 & 5
  - o Species profiles  $(C_2H_4, C_2H_2, C_6H_6) Ref 2 \& 3$
  - $\circ$  Species profiles (CO, CO<sub>2</sub>, CH<sub>4</sub>, C<sub>2</sub>H<sub>2</sub>, C<sub>3</sub>H<sub>4</sub>, C<sub>3</sub>H<sub>6</sub>, C<sub>4</sub>H<sub>2</sub>, C<sub>4</sub>H<sub>4</sub>, C<sub>4</sub>H<sub>6</sub>, C<sub>5</sub>H<sub>6</sub>, C<sub>6</sub>H<sub>6</sub>) **Ref 5**

- PAH profiles (A2, A2R5, A3, A4, A4R5, FLTN) Ref 3 & 4
- o Soot profiles Ref 2, 5, 7
- H/C ratio Ref 6 & 7
- $V_0=6$ cm/s Ref 2
  - o Temperature profile
  - o Species profiles (C2H4, C2H2, C6H6) Ref 2 & 3
  - o PAH profiles (A2, A2R5, A3, A4, A4R5, FLTN) Ref 3
  - Soot profiles

φ=3.0 (C/O=1.0) – Fuel: 50.0 % - O2: 50.0 %

- $V_0 = 4 \text{cm/s}$ **Ref 5** 
  - o Temperature profile
  - Species profiles (CO, CO<sub>2</sub>, CH<sub>4</sub>, C<sub>2</sub>H<sub>2</sub>, C<sub>3</sub>H<sub>4</sub>, C<sub>3</sub>H<sub>6</sub>, C<sub>4</sub>H<sub>2</sub>, C<sub>4</sub>H<sub>4</sub>, C<sub>4</sub>H<sub>6</sub>, C<sub>5</sub>H<sub>6</sub>, C<sub>6</sub>H<sub>6</sub>)
  - Total PAH weight profiles
  - Soot profiles

## **Notes**

Ref 5 lists slightly different equivalence ratios than the other refs, namely  $\phi$ =2.42 and  $\phi$ =3.03 *vs*.  $\phi$ =2.4 and  $\phi$ =3.0.

The mole fractions are reported "as measured", that means as "dry values"; the water was removed from sampled gases before the measures.

Some of the species concentrations are given with units of g of the species per cm<sup>3</sup> at STP conditions (0°C,  $10^5$  Pa) of sampled gases.

## References

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